

This article was downloaded by:

On: 27 January 2011

Access details: *Access Details: Free Access*

Publisher *Taylor & Francis*

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Phosphorus, Sulfur, and Silicon and the Related Elements

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t713618290>

Reaction of Hydrazonoyl Halides 51¹: A Facile Synthesis of 5-Arylthiazoles and Triazolino[4,3-*a*]pyrimidines as Antimicrobial Agents

Abdou O. Abdelhamid^a; Abdelwahed R. Sayed^b; Yasser H. Zaki^b

^a Department of Chemistry, Faculty of Science, Cairo University, Giza, Egypt ^b Department of Chemistry, Faculty of Science, Beni-Suef University, Beni-Suef, Egypt

To cite this Article Abdelhamid, Abdou O. , Sayed, Abdelwahed R. and Zaki, Yasser H.(2007) 'Reaction of Hydrazonoyl Halides 51¹: A Facile Synthesis of 5-Arylthiazoles and Triazolino[4,3-*a*]pyrimidines as Antimicrobial Agents', *Phosphorus, Sulfur, and Silicon and the Related Elements*, 182: 7, 1447 — 1457

To link to this Article: DOI: 10.1080/10426500701242145

URL: <http://dx.doi.org/10.1080/10426500701242145>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.informaworld.com/terms-and-conditions-of-access.pdf>

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

Reaction of Hydrazonoyl Halides 51¹: A Facile Synthesis of 5-Arylthiazoles and Triazolino[4,3-*a*]pyrimidines as Antimicrobial Agents

Abdou O. Abdelhamid

Department of Chemistry, Faculty of Science, Cairo University,
Giza, Egypt

Abdelwahed R. Sayed

Yasser H. Zaki

Department of Chemistry, Faculty of Science, Beni-Suef University,
Beni-Suef, Egypt

*[5-Substituted 2-(3-phenyl-5-substituted 2-pyrazolinyl)(1,3-thiazol-4-yl)]phenyl-diazenes, triazolo[3,4-*a*]pyrimidines, and 2,3-dihydro-1,3,4-thiadiazoles were synthesized with good yields from reactions of hydrazonoyl halides with 5-substituted-3-phenyl-4,5-dihydropyrazole-1-carboximidothionic acid, pyrimidine-2-thione, methyl carbodithioate, respectively. All structures of the newly synthesized compounds were elucidated by elemental analysis, spectral data, and alternative synthesis methods. Newly developed compounds are capable of inhibiting the growth of bacteria (gram positive and gram negative) greatly.*

Keywords 1,3,4-Thiadiazolines; aryazothiazoles; hydrazonoyl halides; triazolino[4,3-*a*]pyrimidines

INTRODUCTION

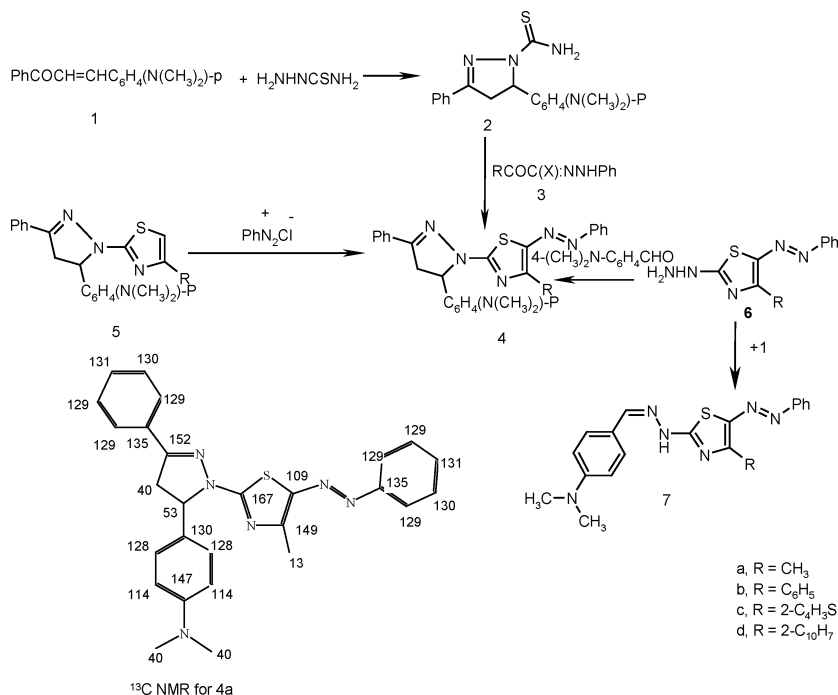
Variously substituted pyrazolines and their derivatives are important biological agents, and a significant amount of research activity has been directed towards this class. In particular, they are used as antitumor,² antibacterial, antifungal, antiviral, antiparasitic, antitubercular, and insecticidal agents.^{3–11} Some of these compounds also have anti-inflammatory, antidiabetic, anaesthetic, and analgesic properties.^{12–15} Moreover, pyrazololines have played a crucial part in the development of theory in heterocyclic chemistry.^{16–20} In this article, we report the reactivity of α -keto hydrazonoyl halides towards

Address correspondence to Abdou Abdelhamid, Department of Chemistry, Faculty of Science, Cairo University, Giza 12316, Egypt. E-mail: abdou_abdelhamid@yahoo.com

4,5-dihydropyrazole-1-carboximidothionic acid, methyl carbodithioate, and 1,3-dihydropyrimidine-2-thione derivatives.

RESULTS AND DISCUSSION

3-(4-Dimethylaminophenyl)-1-phenylpropenone (**1**) was reacted with thiosemicarbazide in ethanol containing catalytically amount of triethylamine gave 5-(4-dimethylaminophenyl)-3-phenyl-4,5-dihydropyrazol-1-carbothioic acid amide (**2**). Structure **2** was elucidated by elemental analysis, spectra, and chemical transformation. ^1H NMR spectra showed signal at $\delta = 2.81$ (s, 6H, $\text{N}(\text{CH}_3)_2$), 3.26 (dd, $J = 15.0, 8.2$ Hz, 1H, $\text{CH}_{2(\text{pyraz})}$), 3.82 (dd, $J = 15.0, 12.2$ Hz, 1H, $\text{CH}_{(\text{pyraz})}$), 5.36 (br, s, 2H, NH_2), 5.68 (dd, $J = 12.2, 8.2$ Hz, 1H, $\text{CH}_{2(\text{pyraz})}$), 7.21–7.81 (m, 9H, ArH's). Thus, compound **2** was reacted with *C*-acetyl-*N*-phenylhydrazonoyl chloride (**3a**) in boiling ethanolic triethylamine gave dimethyl-4-[2-(5-methyl-4-phenylazothiazo-2-yl)-5-phenyl-3,4-dihydro-2H-pyrazol-3-yl]phenylamine (**4a**) (Scheme 1).



SCHEME 1

Structure **3** was elucidated by elemental analysis, spectra and alternative synthesis route. ^1H NMR spectrum of **4a** showed signals δ = 2.47, (s, 3H, CH_3), 2.81 (s, 6H, $\text{N}(\text{CH}_3)_2$), 3.26 (dd, J = 15.0, 8.2 Hz, 1H, $\text{CH}_2(\text{pyraz})$), 3.82 (dd, J = 15.0, 12.2 Hz, 1H, $\text{CH}(\text{pyraz})$), 5.68 (dd, J = 12.2, 8.2 Hz, 1H, $\text{CH}_2(\text{pyraz})$), 7.21–7.81 (m, 14H, ArH's). ^{13}C NMR showed signals agreement with its structure (Scheme 1). Also, treatment of **6a**¹ with **1** gave product identical in all respects (m.p., mixed m.p., and spectra) with **4a**. Also, **5a** was reacted with benzenediazonium chloride in ethanolic sodium acetate solution afforded **4a**. Meanwhile, the appropriate **6a–c** was reacted with 4-(*N,N*-dimethyl)aminobenzaldehyde gave the corresponding thisemicarbazone **7a–c**, respectively.

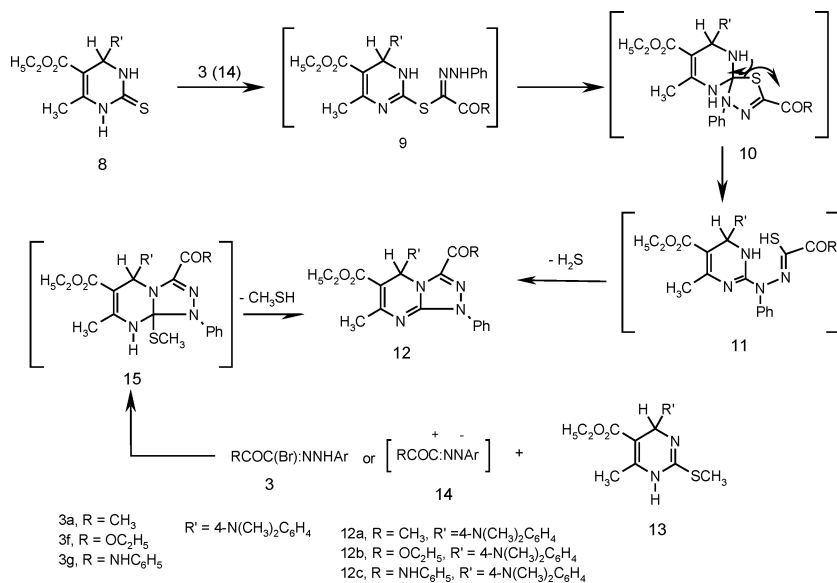
Treatment of ethyl 6-[4-(dimethylamino)phenyl]-4-methyl-2-thioxo-1,3,6-trihydropyrimidine-5-carboxylate (**8**) with the appropriate hydrasonoyl halides **3a–c** in boiling chloroform under reflux gave triazolino[4,3-*a*]pyrimidines **12a–c**, respectively.

Structure of **12** was elucidated on the basis of elemental analysis, spectral data and alternative synthesis route. Thus, treatment of **3a** with ethyl 4-[4-(dimethylamino)phenyl]-6-methyl-2-methylthio-3,4-dihydropyrimidine-5-carboxylate (**13**) in boiling ethanolic sodium ethoxide solution under reflux afforded product identical in all respects (m.p., mixed m.p., and spectra) with **12a**.

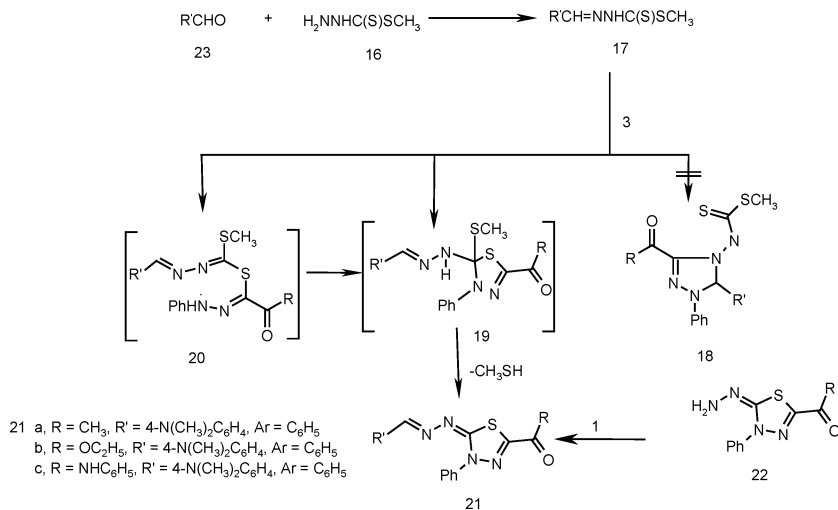
Two possible pathways can account for the formation **12** via 1,3-addition of the thiol tautomer of **8** to the nitrilium imide **14**, (which prepared insitu from hydrasonoyl halide and triethylamine), can give the thiohydrasonate ester **9**, which undergoes nucleophilic cyclization to yield spiro compounds **10** or 1,3-cycloaddition of nitrilium imide **14** to C=S double bond of **8** can give directly **10**. The latter intermediate **10** was converted to **11**, which cyclized immediately to yield **12** by loss hydrogen sulfide (Scheme 2).

On the other hand, treatment of *N,N*-dimethylbenzaldehyde with methyl hydrazinecarbodithioate²¹ (**16**) in 2-propanol afforded (1-aza-2-[4-(dimethylamino)-phenyl]vinylamino)methylthiomethane-1-thione (**17**) (Scheme 3).

Structure **17** was elucidated by elemental analysis, spectral analysis, and chemical transformation. Compound **17** reacted with **3b** in ethanolic triethylamine at room temperature to give one isolate product formulated as: ethyl 2-1,2-diaza-3-[4-(dimethylamino)phenyl]prop-2-enylidene-3-phenyl-1,3,4-thiadiazoline-5-carboxylate (**21b**). Structure **21** was confirmed by elemental analysis and spectral data alternative synthesis route. Thus, 2,3-dihydro-1,3,4-thiadiazole **22**²² was reacted with 4-*N,N*-dimethylbenzaldehyde (**23**) in ethanol afforded product identical in all respects (m.p., mixed m.p. and spectra) with **21b**.



SCHEME 2



SCHEME 3

Antimicrobial Activity

The tested microorganisms were gram +ve bacteria, gram -ve bacteria, and some Fungal-plant. Sensitivity of the selected microorganisms

TABLE I Response of Various Microorganisms to Some Synthesized Compounds In Vitro (Culture)

Microorganisms/ Compound no.	<i>Bacillus subtilis</i> (G ⁺)	<i>Echerichia coli</i> (G ⁻)	<i>Staphylococcus albus</i> (G ⁺)	<i>Streptococcus faecalis</i> (G ⁺)	<i>Candida albicans</i> (Fungus)	<i>Aspergillus flvus</i> (Fungus)
Ampicillin/ Tetracycline	33/30	39/34	34R / 27	37/31	20/37	0.0/0.0
4b	14	14	13	15	13	0.0
4c	14	14	14	14	13	0.0
4d	14	15	15	15	14	0.0
7a	15	14	15	14	15	0.0
7b	13	13	15	14	13	0.0
7c	13	13	14	15	13	0.0
7d	14	14	13	14	15	0.0
12a	15	15	15	14	16	12
12b	16	15	15	16	16	0.0
12c	15	15	14	15	13	0.0
17	14	15	15	14	14	0.0
21a	14	13	14	13	13	0.0
21b	15	14	14	14	14	0.0
21c	13	14	13	13	13	0.0

St. Reference standard; Chloramphenicol was used as a slandered antibacterial agent; Terbinafin was used as a slandered antifungal agent. Values show zone of inhibition in mm. Diameter of the inhibition zones were: high (11–15 mm), moderate (6–10 mm), slight (1–5 mm), and negative (0).

to some synthesized compounds were determined by in vitro culture that were dissolved in chloroform; the tests were carried out using the filter paper and hole plate method.^{23,24} Studies on the biological activity of compounds in comparison with Ampicillin and tetracycline showed in Table 1. In general all tested compounds were capable of a high inhibiting the growth of gram positive and gram negative. Also, all compounds were a high inhibition towards *Candida albicans* and negative *Aspergillus flvus* (except **12a** is high).

EXPERIMENTAL

All melting points were determined on an electrothermal apparatus and are uncorrected. IR spectra were recorded (KBr discs) on a Shimadzu FT-IR 8201 PC spectrophotometer. ¹H NMR and ¹³C NMR spectra were recorded in CDCl₃ and (CD₃)₂SO solutions on a Varian Gemini 300 MHz spectrometer, and chemical shifts are expressed in δ units using TMS as an internal reference. Elemental analyses and microorganism tests were carried out at the Microanalytical Center of the Cairo University. Hydrazonoyl halides were prepared as previously reported.^{25–30}

5-(4-Dimethylaminophenyl)-3-phenyl-4,5-dihydropyrazol-1-carbothioic Acid Amide (**2**)

A mixture of 3-(4-dimethylaminophenyl)-1-phenylpropenone (**1**) (1.25 g, 5 mmol), thiosemicarbazide (0.045 g, 5 mmol) in ethanol (20 mL) and triethylamine (2 drops) were boiled under reflux for 2 h. The resulting solid, which formed after cooling, was collected and crystallized from ethanol to give **2** (Tables II and III).

Dimethyl-4-[2-(5-substituted 4-phenylazothiazo-2-yl)-5-phenyl-3,4-dihydro-2H-pyrazol-3-yl]phenylamine (**4a-d**)

A mixture of the appropriate hydrazonoyl halides (**3a-d**) (5 mmol), 5-(4-dimethylaminophenyl)-3-phenyl-4,5-dihydropyrazol-1-carbothioic acid amide (**2**) (1.67 g, 5 mmol), triethylamine (0.75 mL, 5 mmol) in ethanol (20 mL) were boiled under reflux for 3 h. The resulting solid was collected and recrystallized from acetic acid to give **4a-d**, respectively (Tables II and III).

Synthesis of (**4a**): Alternative Methods

Method (A)

A mixture of (4-methyl-5-phenylazothiazol-2-yl)hydrazine (**6a**)¹ (1.97, 5 mmol) and 3-(4-dimethylaminophenyl)-1-phenylpropenone (**1**) (1.25 g, 5 mmol) in ethanol (20 mL) containing catalytic amount of piperidine (2 drops) was boiled under reflux for 2 h to give **4a**.

Method (B)

Benzenediazonium chloride (5 mmol), which prepared from aniline (0.45 mL, 5 mmol), hydrochloric acid (6 N, 6 mL), and sodium nitrite (0.35g, 5 mmol), was added dropwise with stirring to a cold solution of a mixture of dimethyl-4-[1-(4-methyl)(1,3-thiazol-2-yl))-3-phenyl(2-pyrazolin-5-yl)]phenylamine (**5**) (1.81 g, 5 mmol) and sodium acetate trihydrate (1.3 g, 10 mmol) in ethanol (50 mL). The resulting solid was collected and recrystallized to give product identical with **4a**.

Synthesis of Dimethyl-4-[1-(4-methyl)(1,3-thiazol-2-yl))-3-phenyl(2-pyrazolin-5-yl)]phenylamine (**5**)

A mixture of 5-(4-dimethylaminophenyl)-3-phenyl-4,5-dihydropyrazol-1-carbothioic acid amide (**2**) (1.67 g, 5 mmol) and chloroacetone (0.46 g, 5 mmol) in ethanol (20 mL) were heated under reflux for 2 h. The resulting solid was collected and recrystallized from ethanol to give **5** (Tables II and III).

TABLE II Characterization Data of the Newly Synthesized Compounds

Compd. no.	M.p., °C solvent	Yield % colour	Mol. formula mol. wt.	% Analyses		calcd./found	
				C	H	N	S
2	198–200 EtOH	83 Yellow	C ₁₈ H ₂₀ N ₄ S 324.45	66.64 66.42	6.21 6.32	17.27 17.20	9.88 10.00
4a	165–66 DMF/EtOH	80 Deep red	C ₂₇ H ₂₆ N ₆ S 466.61	69.50 69.60	5.62 5.42	18.01 17.92	6.87 6.70
4b	194–95 DMF/EtOH	78 Deep red	C ₃₂ H ₂₈ N ₆ S 528.68	72.70 72.92	5.34 5.43	15.90 15.75	6.06 5.85
4c	188–90 DMF/EtOH	68 Deep red	C ₃₀ H ₂₆ N ₆ S ₂ 534.71	67.39 67.52	4.90 4.80	15.72 15.60	11.99 12.12
4d	165–57 DMF/EtOH	74 Deep red	C ₃₆ H ₃₀ N ₆ S 578.74	74.71 74.62	5.22 5.10	14.52 14.72	5.54 5.42
5	210–12 EtOH	86 Yellow	C ₂₁ H ₂₂ N ₄ S 362.50	69.58 69.65	6.12 5.95	15.46 15.62	8.85 8.64
7a	190–91 Dioxan	88 Red	C ₁₉ H ₂₀ N ₆ S 364.48	62.61 62.54	5.53 5.35	23.06 22.85	8.80 8.65
7b	194–95 Dioxan	75 Red	C ₂₄ H ₂₂ N ₆ S 426.55	67.58 67.70	5.20 5.12	19.70 19.61	7.52 7.45
7c	149–51 Dioxan	78 Deep red	C ₂₂ H ₂₀ N ₆ S ₂ 432.57	61.09 61.20	4.66 4.55	19.43 19.34	14.82 14.72
7d	195–61 Dioxan	83 Deep red	C ₂₈ H ₂₄ N ₆ S 476.61	70.56 70.65	5.08 4.89	17.63 17.50	6.73 6.80
8	200–202 EtOH	72 Yellow	C ₁₆ H ₂₁ N ₃ O ₂ S 319.43	60.16 60.30	6.63 6.53	13.15 13.23	10.04 10.12
12a	185–87 EtOH	82 Yellow	C ₂₅ H ₂₇ N ₅ O ₂ 445.53	67.40 67.50	6.11 6.18	15.72 15.85	— —
12b	165–66 EtOH	78 Yellow	C ₂₆ H ₂₉ N ₅ O ₄ 475.55	65.67 65.70	6.15 5.84	14.73 14.90	— —
12c	150–521 EtOH	82 Yellow	C ₃₀ H ₃₀ N ₆ O ₃ 522.61	68.95 68.80	5.79 5.90	16.08 15.88	— —
13	138–40 EtOH	85 Yellow	C ₁₇ H ₂₃ N ₃ O ₂ S 333.46	61.23 61.12	6.95 6.80	12.60 12.48	9.62 9.71
17	190–91 EtOH	82 Yellow	C ₁₁ H ₁₅ N ₃ S ₂ 253.39	52.14 52.00	5.97 5.85	16.58 15.72	25.31 25.22
21a	148–49 EtOH	74 Dioxan	C ₁₉ H ₁₉ N ₅ OS 365.46	62.45 62.55	5.24 5.40	19.16 19.30	8.77 8.85
21b	200–202 EtOH	77 Dioxan	C ₂₀ H ₂₁ N ₅ O ₂ S 395.49	60.74 60.68	5.53 5.35	17.71 17.90	8.11 8.20
21c	205–207 EtOH	71 Dioxan	C ₂₄ H ₂₂ N ₆ OS 442.55	65.14 65.26	5.01 5.00	18.99 19.12	7.25 7.30

[4-(2-Aza-2-[4-substituted 5-phenyldiazenyl](1,3-thiazol-2-yl)]aminovinyl)phenyl]-dimethylamine 7a–c

A mixture of 4-(N,N-dimethylamino)benzaldehyde (**23**) (0.74a, mmol), the appropriate **6a–c** (5 mmol) and acetic acid (2 drops) in ethanol (20

TABLE III Spectra of Some Newly Synthesized Compounds

Comp. no.	Spectral data
2	¹ H NMR: 2.81 (s, 6H, N(CH ₃) ₂), 3.26 (dd, J = 15.0, 8.2 Hz, 1H, CH ₂ (pyraz)), 3.82 (dd, J = 15.0, 12.2 Hz, 1H, CH ₂ (pyraz)), 5.36 (br, s, 2H, NH ₂), 5.68 (dd, J = 12.2, 8.2 Hz, 1H, CH ₂ (pyraz)), 7.21–7.81 (m, 9H, ArH's).
4a	¹ H NMR: 2.47, (s, 3H, CH ₃), 2.81 (s, 6H, N(CH ₃) ₂), 3.26 (dd, J = 15.0, 8.2 Hz, 1H, CH ₂ (pyraz)), 3.82 (dd, J = 15.0, 12.2 Hz, 1H, CH ₂ (pyraz)), 5.68 (dd, J = 12.2, 8.2 Hz, 1H, CH ₂ (pyraz)), 7.21–7.81 (m, 14H, ArH's).
4b	¹ H NMR: 2.81 (s, 6H, N(CH ₃) ₂), 3.26 (dd, J = 15.0, 8.2 Hz, 1H, CH ₂ (pyraz)), 3.82 (dd, J = 15.0, 12.2 Hz, 1H, CH ₂ (pyraz)), 5.68 (dd, J = 12.2, 8.2 Hz, 1H, CH ₂ (pyraz)), 7.21–7.81 (m, 19H, ArH's).
4c	¹ H NMR: 2.81 (s, 6H, N(CH ₃) ₂), 3.26 (dd, J = 15.0, 8.2 Hz, 1H, CH ₂ (pyraz)), 3.82 (dd, J = 15.0, 12.2 Hz, 1H, CH ₂ (pyraz)), 5.36 (br, s, 2H, NH ₂), 5.68 (dd, J = 12.2, 8.2 Hz, 1H, CH ₂ (pyraz)), 7.21–7.81 (m, 17H, ArH's).
4d	¹ H NMR: 2.81 (s, 6H, N(CH ₃) ₂), 3.26 (dd, J = 15.0, 8.2 Hz, 1H, CH ₂ (pyraz)), 3.82 (dd, J = 15.0, 12.2 Hz, 1H, CH ₂ (pyraz)), 5.36 (br, s, 2H, NH ₂), 5.68 (dd, J = 12.2, 8.2 Hz, 1H, CH ₂ (pyraz)), 7.21–7.81 (m, 21H, ArH's).
5	¹ H NMR: 2.47, (s, 3H, CH ₃), 2.81 (s, 6H, N(CH ₃) ₂), 3.26 (dd, J = 15.0, 8.2 Hz, 1H, CH ₂ (pyraz)), 3.82 (dd, J = 15.0, 12.2 Hz, 1H, CH ₂ (pyraz)), 5.68 (dd, J = 12.2, 8.2 Hz, 1H, CH ₂ (pyraz)), 7.21–7.81 (m, 14H, ArH's).
7a	¹ H NMR: 2.21(t, 3H, CH ₃), 2.85 (s, 6H, N(CH ₃) ₂), 6.46-7.48 (m, 9H, ArH's), 8.21(s, 1H, CH=N), 9.24 (s, br., 1H, NH).
7b	2.85 (s, 6H, N(CH ₃) ₂), 6.46-7.48 (m, 14H, ArH's), 8.21(s, 1H, CH=N), 9.24 (s, br., 1H, NH).
7c	¹ H NMR: 2.85 (s, 6H, N(CH ₃) ₂), 6.46-7.48 (m, 12H, ArH's), 8.21(s, 1H, CH=N), 9.24 (s, br., 1H, NH).
7d	¹ H NMR: 2.85 (s, 6H, N(CH ₃) ₂), 6.46-7.89 (m, 19H, ArH's), 8.21(s, 1H, CH=N), 9.24 (s, br., 1H, NH).
8	¹ H NMR: 1.30 (t, 3H, CH ₂ CH ₃), 1.74 (s, 3H, CH ₃), 2.85 (s, 6H, N(CH ₃) ₂), 4.12 (q, 2H, CH ₂ CH ₃), 4.59 (s, 1H, pyrimidine C4), 6.46-6.88 (m, 4H, ArH's), 9.45 (s, br., 2H, 2NH).
12a	¹ H NMR: 1.31 (t, 3H, CH ₂ CH ₃), 1.71 (s, 3H, CH ₃), 2.0 (s, 3H, CH ₃), 2.85 (s, 6H, N(CH ₃) ₂), 4.19 (q, 2H, CH ₂ CH ₃), 4.59 (s, 1H, pyrimidine C4), 6.46-7.04 (m, 9H, ArH's). IR: 3056, 2984 (CH), 1713, 1665 (CO's), 1620 (C=N) and 1600 (C=C).
12b	¹ H NMR: 1.31 (t, 3H, CH ₂ CH ₃), 1.29 (t, 3H, CH ₂ CH ₃), 1.71 (s, 3H, CH ₃), 2.85 (s, 6H, N(CH ₃) ₂), 4.19 (q, 2H, CH ₂ CH ₃), 4.22 (q, 2H, CH ₂ CH ₃), 4.59 (s, 1H, pyrimidine C4), 6.46-7.05 (m, 9H, ArH's). IR: 3060, 2975 (CH), 1713 (CO's), 1625 (C=N) and 1610 (C=C).
12c	¹ H NMR: 1.31 (t, 3H, CH ₂ CH ₃), 1.71 (s, 3H, CH ₃), 2.85 (s, 6H, N(CH ₃) ₂), 4.19 (q, 2H, CH ₂ CH ₃), 4.59 (s, 1H, pyrimidine C4), 6.46-6.88 (m, 4H, ArH's), 9.45 (s, br., 1H, 1NH). IR: 3065, 2989 (CH), 1713, 1680 (CO's), 1620 (C=N) and 1605 (C=C).
13	¹ H NMR: 1.31 (t, 3H, CH ₂ CH ₃), 1.71 (s, 3H, CH ₃), 2.0 (s, 3H, CH ₃), 2.85 (s, 6H, N(CH ₃) ₂), 4.19 (q, 2H, CH ₂ CH ₃), 4.59 (s, 1H, pyrimidine C4), 6.46-6.88 (m, 4H, ArH's), 9.45 (s, br., 1H, 2NH).
17	¹ H NMR: 2.12 (s, 3H, CH ₃), 2.85 (s, 6H, N(CH ₃) ₂), 6.61–7.42 (m, 9H, ArH's), 8.45 (s, 1H, CH=N), 9.23 (s, br., 1H, NH).
21a	¹ H NMR: 1.31 (t, 3H, CH ₂ CH ₃), 2.85 (s, 6H, N(CH ₃) ₂), 4.22 (q, 2H, CH ₂ CH ₃), 6.46–7.48 (m, 9H, ArH's), 8.45 (s, 1H, CH=N) IR: 1665 (CO), 1620 (C=N) and 1600 (C=C).
21b	¹ H NMR: 2.20 (t, 3H, CH ₃), 2.85 (s, 6H, N(CH ₃) ₂), 6.46–7.48 (m, 8H, ArH's), 8.45 (s, 1H, CH=N). IR: 1713 (CO), 1625 (C=N) and 1600 (C=C).
21c	¹ H NMR: 2.85 (s, 6H, N(CH ₃) ₂), 6.46–7.65 (m, 14H, ArH's), 8.45 (s, 1H, CH=N), 9.35 (s, br., 1H, NH). IR: 1685 (CO), 1620 (C=N) and 1600 (C=C).

mL) was stirred for 1 h. The resulting solid was collected and recrystallized from dioxan to give **7a–c**, respectively in a good yield (Tables II and III).

Ethyl 6-[4-(Dimethylamino)phenyl]-4-methyl-2-thioxo-1,3,6-trihydropyrimidine-5-carboxylate (8)

A mixture of 4-dimethylaminobenzaldehyde (7.45 g, 50 mmol), ethyl 3-oxobutanoate (6.50 g, 50 mmol) and thiourea (3.8 g, 50 mmol) in ethanol (25 mL) contain hydrochloric acid (1 mL, 12 N) were boiled under reflux for 6 h. The resulting solid, which formed after cooling, was collected and recrystallized from ethanol to give **8** (Tables II and III).

Ethyl 4-[4-(Dimethylamino)phenyl]-6-methyl-1-phenyl-3-(1-acyl)-4,3a-dihydro-1,2,4-triazololino[4,3-a]pyrimidine-5-carboxylate 12a–c

Method A

A mixture of the appropriate hydrazonoyl halides **3a–c** (5 mmol) and **8** (1.59 g, 5 mmol) in chloroform (20 mL) containing triethylamine (0.5 g (0.75 mL), 5 mmol) was refluxed for 10 h. Chloroform was evaporated under reduced pressure and the residue solid was crystallized from ethanol to give ethanol to give **12a–c** (Tables II and III).

Method A

A mixture of the appropriate hydrazonoyl halides **3a–c** (5 mmol), **13** (1.66 g, 5 mmol), and sodium ethoxide (0.34 g, 5 mmol) in ethanol (20 mL) was refluxed for 3 hrs. The reaction mixture was cooled and the resulting solid was collected and crystallized from dioxane gave products identical in all respects (m.p., mixed m.p., and spectra) with corresponding products obtained by Method A.

Ethyl 4-[4-(Dimethylamino)phenyl]-6-methyl-2-methylthio-3,4-dihydropyrimidine-5-carboxylate (13)

A mixture of **8** (1.95 g, 5 mmol) and sodium methoxide (0.29 g, 5 mmol) in methanol (20 mL) were stirred for 3 hrs. Iodomethane (0.72 g, 5 mmol) was added while stirring; after 2 h the resulting solid was collected and recrystallized from ethanol to give **13** (Tables II and III).

(1-Aza-2-[4-(dimethylamino)phenyl]-vinylamino)methylthiomethane-1-thione (17)

A mixture of the 4-dimethylaminobenzaldehyde (1.49 g, 10 mmol) and methyl hydrazinecarbodithioates **16**³¹ (1.22 g, 10 mmol) in 2-propanol (10 mL) were stirred for 2 h at room temperature. The resulting solid was collected and crystallized from ethanol to give yellow crystals **17** (Tables II and III).

5-Acyl-2-1,2-diaza-3-[4-(dimethylamino)phenyl]prop-2-enylidene-3-phenyl-1,3,4-thiadiazolines **21a-c**

A mixture of the methyl carbodithioate **17** (1.26 g, 5 mmol), the appropriate hydrazoneyl halides **3a-c** (5 mmoles), and triethylamine (0.75 mL, 0.005 mol) in ethanol (20 mL) was stirred for 2 h at room temperature. The resulting solid was collected and recrystallized to give 2,3-dihydro-1,3,4-thiadiazoles **21a-c**, respectively (Tables II and III).

Alternative Method

A mixture of ethyl 2-hydrazino-3-phenyl-1,3,4-thiadiazoline-5-carboxylate (**22**) (1.32 g, 5 mmol) and 4-dimethylaminobenzaldehyde (0.74 g, 5 mmol) in 2-propanol (10 mL) were stirred for 2 h at room temperature. The resulting solids were collected and recrystallized from ethanol to give **21a** (Tables II and III).

REFERENCES

- [1] Part 50: A. O. Abdelhamid, A. R. Sayed, and A. M. Elzanati, *Chemistry An Indian J.*, **2**, 287 (2005).
- [2] E. C. Taylor, H. Patel, and H. Kumr, *Tetrahedron*, **48**, 8089 (1992).
- [3] S. G. Roelvan, C. Arnold, and K. Wellnga, *J. Agric. Food Chem.*, **84**, 406 (1979).
- [4] G. H. Keats, Brit. Patent. 1, **209**, 631 (1970).
- [5] R. M. Kedar, N. N. Vidhale, and M. M. Chincholkar, *Orient J. Chem.*, **13**, 143 (1997).
- [6] A. Singh, S. Rathod, B. N. Berad, S. D. Patil, and A. G. Dosh, *Orient J. Chem.*, **16**, 315 (2000).
- [7] H. Z. Katri and S. A. Vunii, *J. Ind. Chem. Soc.*, **58**, 168 (1981).
- [8] N. B. Das and A. S. Mittra, *Ind. J. Chem.*, **16B**, 638 (1978).
- [9] D. Azarifar and M. Shaebanzadeh, *Molecules*, **7**, 885 (2002).
- [10] B. Shivarama Holla, P. M. Akberali, and M. K. Shivanada, *Farmaco*, **55**, 256 (2000).
- [11] E. Palaska, M. Aytimir, I. Tayfin, K. Erol, and E. Dilek, *Eur. J. Med. Chem. Chim. Ther.*, **36**, 539 (2001).
- [12] H. G. Garge, P. Chandra, *J. Pharm. Sc.*, **14**, 649 (1971).
- [13] H. A. Regaila, A. K. El-Bayouk, and M. Hammad, *Egypt J. Chem.*, **20**, 197 (1979).
- [14] R. Krishna, B. R. Pande, S. P. Bhatthwal, and S. S. Parmar, *J. Med. Chem.*, **15**, 567 (1980).
- [15] M. I. Husain and S. Shukla, *Ind. J. Chem.*, **25B**, 983 (1986).

- [16] Yu. V. Tomilovi, G. P. Okonnishnikova, E. V. Shulishov, and O. M. Nefedov, *Russ. Chem. Bt.*, **44**, 2114 (1995).
- [17] E. I. Klimova, M. Marcos, T. B. Klimova, A. T. Cecilio, A. T. Ruben, and R. R. Lena, *J. Organomet. Chem.*, **585**, 106 (1999).
- [18] D. Bhaskarreddy, A. Padmaja, P. V. Ramanareddy, and B. Seenaiiah, *Sulfur Lett.*, **16**, 227 (1993).
- [19] V. Padmavathi, R. P. Sumathi, B. N. Chandrasekhar, and D. Bhaskarreddy, *J. Chem. Research*, **10**, 610 (1999).
- [20] D. Bhaskarreddy, B. Chandrasekhar, V. Padmavathi, and R. P. Sumathi, *Synthesis* **5**, 491 (1998).
- [21] D. L. Klayman, J. F. Bartosevich, T. S. Griffin, C. J. Mason, and J. P. Scovill, *J. Med. Chem.*, **22**, 855 (1979).
- [22] A. O. Abdelhamid, H. F. Zohdi, and N. M. Rateb, *J. Chem. Res. (S)*, **184**, 199. *J. Chem. Res. (M)*, **3**, 920 (1999).
- [23] C. Refer Lefert, H. Siripumchidbouree, S. Hamspons, S. Workman, D. Sigee, H. A. S. Epton, and A. Harbour, *J. Appl. Bact.*, **78**, 97 (1955).
- [24] W. E. Solomons and N. J. Doorenbos, *J. Pharm. Sci.*, **63**, 19 (1974).
- [25] G. Fravel, *Bull. Soc. Chim. Fr.*, **31**, 150 (1904).
- [26] N. E. Eweiss and A. Osman, *Tetrahedron Lett.*, **1169** (1979).
- [27] A. S. Shawali and A. O. Abdelhamid, *Bull. Chem. Soc. Jpn.*, **49**, 321 (1976).
- [28] A. S. Shawali and A. Osman, *Tetrahedron*, **27**, 2517 (1971).
- [29] A. O. Abdelhamid and F. H. H. El-Shiatey, *Phosphorus, Sulfur, Silicon and the Related Elements*, **39**, 45 (1988).
- [30] H. M. Hassaneen, A. S. Shawali, N. M. Elwan, and N. M. Abounada, *Sulfur Letters*, **14**, 41 (1992).